

## Common thermodynamic constants

---

Parameter		Value	Unit
specific gas constant of dry air:	$R_d =$	287	$Jkg^{-1}K^{-1}$
specific gas constant of water vapor:	$R_v =$	461.5	$Jkg^{-1}K^{-1}$
isobaric specific heat of dry air:	$c_d =$	1004	$Jkg^{-1}K^{-1}$
isobaric specific heat of water vapor:	$c_v =$	1884	$Jkg^{-1}K^{-1}$
specific heat of water (liquid):	$c_l =$	4220	$Jkg^{-1}K^{-1}$
specific heat of ice:	$c_i =$	2097	$Jkg^{-1}K^{-1}$
latent heat of vaporization at $0^\circ C$ :	$L_{v,0} =$	$2.5 \cdot 10^6$	$Jkg^{-1}$
latent heat of fusion at $0^\circ C$ :	$L_{f,0} =$	$3.35 \cdot 10^5$	$Jkg^{-1}$
saturation vapor pressure at $0^\circ C$ :	$e_{s,0} =$	611.21	$Pa$

---